Course description (syllabus) form for higher education, doctoral, postgraduate and skills development programmes

A. General course description

[Filled out by teacher/course coordinator except for the following fields marked in blue:

- 1) course title,
- 2) unit organising the course,
- 3) unit for which the course is organised,
- 4) course ID,
- 5) ISCED code,
- 6) ECTS credit allocation,
- 7) form of course completion assessment,
- 8) language of instruction,
- 9) indication whether attempts at obtaining course credit can be repeated,
- 10) affiliation of the course to a course group].

Field name	Comments
Course title	Molekularne podstawy metabolizmu
	Molecular basis of metabolism
Unit organising the course	Faculty of Medicine
	Department of Medical Biology and Biochemistry
	Ludwik Rydygier Collegium Medicum
	Nicolaus Copernicus University
Unit for which the course	Faculty of Medicine
is organised	Field of study: Medicine, full-time studies, long cycle
Course ID	1655-LekM11NPMMPM-J
ISCED code	0912
ECTS credit allocation	4.00
Form of course completion	ungraded credit
assessment	
Language of instruction	English
Indication whether attempts at	по
obtaining course credit can be	
repeated	
Affiliation of the course	
to a course group	
to a course group Total student workload	 1.The workload associated with activities requiring direct participation of academic teachers is: participation in lectures: 30 hours participation in tutorials: 24 hours participation in seminars: 6 hours consultation: 5 hours passing the test: 2 hours The workload associated with classes requiring direct participation of academic teachers is 63 hours, which corresponds to 2.52 ECTS points 2. Balance of student workload: participation in lectures: 30 hours participation in lectures: 4 hours participation in lectures: 5 hours participation in tutorials: 24 hours participation in tutorials: 24 hours participation in tutorials: 5 hours participation in tutorials: 5 hours participation in seminars: 6 hours preparation for tutorials and seminars: 10 hours writing tutorial reports: 5 hours reading the indicated literature: 10 hours

	a angulation 5 hours
	- preparation for passing and passing: 10 hours
	The total student workload is 100 hours, which corresponds to 4 ECTS
	points
	3. Workload related to scientific research:
	- reading of the indicated scientific literature: 10 hours
	- research and scientific consultations: 5 hours
	- participation in lectures (including methodology of scientific research
	- participation in tectures (including methodology of scientific research,
	- participation in tutorials and seminars covered by scientific activity
	(including methodology of scientific research, research results, studies):
	30 hours
	- preparation for tutorials and seminars covered by scientific activity:
	10 hours
	- preparation to pass in the area of research and science aspects for a
	given subject: 10 hours
	- writing reports on tutorials covered by scientific activity: 5 hours
	The total amount of student work associated with the conducted research
	1 ne total amouni of sludeni work associated with the conducted research
	is 100 nours, which corresponds to 4 ECIS points
	4. <i>Time required to prepare and participate in the assessment process:</i>
	- preparation for passing + credit: 10 hours
	- reading the indicated literature: 10 hours
	The total amount of student work involved in preparing for the
	participation in the assessment process is 20 hours, which corresponds
	to 0.2 ECTS points
	5 Balance of student workload of a practical nature
	narticipation in tutorials: 24 hours
	- participation in futorials. 24 hours
	The total student workload of a practical hature is 24 hours, which
	corresponds to 0.96 ECIS points
	6. Time required for compulsory placement:
	not applicable
Learning outcomes –	<i>W1: knows the structure of amino acids and their physical and chemical</i>
knowledge	properties - B K_W10
	W2: knows the structure of carbohydrates and their physical and
	chemical properties - B K_W10
	W3: knows the structure of lipids and their function at the cellular and
	extracellular levels - B K W11
	W4: knows the structure of polysaccharides and their function at the
	cellular and extracellular levels - BK W11
	W5: characterizes I II III and IV structures of proteins BK W12
	W5. characterizes 1-, 11-, 111- and 1V structures of proteins- D K_W12
	we knows the post-translational and junctional modifications of the D_{K} with
	protein and their importance - B K_w12
	W/: characterizes proteins participating in chromatin structure - B
	K_W13
	W8: describes the role of enzymes and vitamins in metabolic processes
	- B K_W15
	W9:knows the regulatory mechanisms of metabolic processes - B
	K_W15
	W10: describes the effect of proper diet. digestion processes and
	absorption on the course of metabolic nathways - $R K W15$
	W11: describes antioxidant function of vitamins - R K W17
	W12: describes the role of blood in maintaining the body's water
	alastrobyta balance BK W01
	electronyle bullance – $D K_WUI$ W12, observations the participation of blood buffers in maintain i
	w15: cnaracterizes the participation of blood buffers in maintaining
	body homeostasis – B K_W02

Learning outcomes – skills	<i>U1: calculates molar and percentage concentrations of diluted solutions - B K_U03</i>
	<i>U2: determines the pH of the solution and the effect of pH changes on</i> <i>the properties of proteins and carbohydrates - B K U05</i>
	U3: uses qualitative analysis, titration, colorimetry, pH-parameters,
	analyzes and interprets the results obtained - B K_U09
	<i>U4: uses analytical balance, spectrophotometer, pH-meter and</i>
Learning outcomes social	evaluates the accuracy of measurements $-BK_U10$
competence	constant supplementing - K K01
Teaching methods	Lecture:
	expository teaching methods - informative (traditional) lecture with multimedia presentation, problem-based teaching methods - discussion method
	seminar:
	expository teaching methods - a talk, problem-based teaching methods
	- classic problem method, discussion method, case method, discovery
	tutorial:
	expository teaching methods - explanation, problem-based teaching methods – classic problem method, didactic discussion, discovery
	teaching methods - laboratory practicals, practical exercises,
	demonstration
Prerequisites	Student beginning the course "Molecular basis of metabolism should possess the knowledge in the field of chemistry and biology at the advanced level of secondary school exam
Brief course description	"Molecular basis of metabolism" course is carried out in the first
*	semester. It consists of 30 hours of lectures, 24 hours of tutorials, and
	6 hours of seminars. The aim of the course is to prepare students for learning clinical subjects, as well as to work in a future profession. The topics provided allow to get to know the structure of the most
	important biomolecules (proteins, carbohydrates, vitamins, lipids),
	their physiological functions, as well as the influence of their
	pathological modifications on the course of human metabolism. Knowledge about the structure of enzymes, their activities and methods
	of regulation is necessary to understand metabolic changes in the states of physiology and pathology.
Full course description	Lectures:
	1. Protein amino acids, their characteristics. Structure and properties of peptide bond. Structure and function of biologically important peptides Biosynthesis and structure of insulin
	2 Proteins - classification, characteristics of I II III- and IV-
	structures. Post-translational modification of proteins. Structure of ribonuclease. Structure and synthesis of collagen.
	3. Relationships between protein structure and its function - myoglobin,
	hemoglobin, prions, immunoglobulins. The physiological role of hemoglobin and myoglobin. The mechanism of oxygen binding with myoglobin and hemoglobin. Changes in the structure of hemoglobin at
	various stages of human development. Glycosylated hemoglobin. Hemoglobinopathies.
	4. Blood functions. Organic and inorganic components of plasma.
	Characteristics and functions of plasma proteins. Buffering properties of blood - carbonate buffer and the role of hemoglobin in buffering (Pohr and Halden effect)
	(Doni una rauaen ejjeci). 5 Enzyme - structure features enzyme-substrate complexes. The
	concept of isoenzyme. Physical and chemical properties of isoenzymes.
	The significance of isoenzymes in diagnostics. Classification of enzymes.

	 6. Michaelis-Menten kinetics of enzymatic reaction and allosteric enzyme kinetics. Regulation of enzyme activity. Types of inhibition with examples of inhibitors used in medicine. 7. Water- and fat-soluble vitamins - structure and importance in metabolic processes. The structure of coenzymes and functions of coenzymes in enzymatic reactions. 8. The role of enzymes in the digestion of carbohydrates, lipids, proteins and nucleic acids. Composition and role of digestive juices. The role of bile acids in the process of digestion. Enterohepatic circulation of bile acids. Molecular mechanisms of absorption of digestive products. 9. Classification of carbohydrates, examples of biologically important mono-, di-, and polysaccharides. Glucosaminoglucans and glycoproteins - structure, examples, functions in the body. 10. Saturated and unsaturated fatty acids, their nomenclature and structure. Simple and complex lipids - examples, structure, function. Cholesterol structure. Cholesterol derivatives and their role in the body. Seminars: Disturbed protein synthesis and structure in the pathogenesis of selected diseases. The importance of enzymes in the diagnosis of selected diseases. Enzymes as a target of pharmacotherapy. Qualitative analysis of amino acids and proteins. Qualitative analysis of proteins. Qualitative and quantitative analysis of selected vitamins.
	 6. Selected properties of digestive juices. 7. Selected properties of monocarbohydrates.
* •.	8. Revision.
Literature	 Leading textbook: Rodwell VW, Bender D, Botham KM, Kennelly PJ, Weil PA. Harpers Illustrated Biochemistry, 30th Edition, McGraw-Hell Mecical 2015 (or 31th Edition, 2018) Supplementary handbooks: Lieberman M, Peet A. Marks' Basic Medical Biochemistry. A Clinical Approach, 5th Edition, Wolters Kluwer 2018 Ferrier DR. Lippincott Illustrated Reviews: Biochemistry, 7th Edition, Wolters Kluwer 2017 Ronner P. Netter's Essential Biochemistry, Elsevier 2018
Assessment methods and criteria	 Intrasemestral test (MCQ): W1, W4-W13, U1-U2, K1 In the case of the intrasemestral test, the points obtained are converted into grades according to the following scale: <u>% points</u> Grade <u>92-100</u> Very good <u>84-91</u> Good plus <u>76-83</u> Good <u>68-75</u> Satisfactory plus <u>56-67</u> Satisfactory <u>0-55</u> Fail Oral or written answer (evaluation of active participation and preparation for the classes): W1-W13, U1-U4, K1. In order to pass the student must get a minimum of 60% of the points from the answer.

	In the case of an oral or written answer to the assessment of the learning outcomes achieved by the student, the following criteria are applied: - credit in the case when: the student knows the basic issues and has mastered the program minimum, understands the questions asked, presents his knowledge in a logical and systematic way, can practically apply the acquired knowledge; - failure to pass if: the student has not mastered the program minimum, does not understand the questions, provides answers not on the subject, does not properly use the basic vocabulary, can not practically apply the acquired knowledge.
	3. Focused observation of the student's activity while performing practical tasks: U1-U4. In order to pass the student must get a minimum of 60% of points for a properly completed activity.
	4. Report: W1-W13, U1-U4, K1. In order to pass the student must get a minimum of 60% of points for the presented report on the tutorials.
	5. Oral presentation: W1-W13, K1. In order to pass the student must get a minimum of 60% of points for the presentation.
	6. Activity – extended observation: K1. In order to pass the student must get a minimum of 50%.
Work placement	not applicable

B. Description of the course within the period of instruction

[Filled out by teacher/course coordinator except for the following fields marked in blue:

- 1) period of instruction,
- 2) form of assessment of course completion in the period of instruction,
- 3) form(s) of classes, number of hours and completion assessment methods,
- 4) name of course coordinator in the period of instruction,
- 5) names of persons managing student groups for the course,
- 6) course attributes,
- 7) course groups including description and limit to the number of students within the groups,

Field name	Komentarz
Period of instruction	Year I, semester I - winter semester
Form of assessment of course completion in the period of instruction	ungraded credit
Form(s) of classes, number of hours and completion assessment methods	Lectures: 30 hours, ungraded credit Seminars: 6 hours, ungraded credit Tutorials: 24 hours, ungraded credit
Name of course coordinator in the period of instruction	Karolina Szewczyk-Golec, PhD, Associate Professor
Names of persons managing student groups for the course	Karolina Szewczyk-Golec, PhD, Daria Kupczyk, PhD, Rafał Bilski, MSc

Course attributes	Obligatory subject
Course groups including	Lectures for the whole year
description and limit to the	Seminar groups up to a maximum of 24 students
number of students within the	Tutorial groups up to a maximum of 12 students
groups	
Time and place of classes	dates and places of classes are specified by the Department of Didactics of Collegium Medicum tutorials and seminars: lab classes at the Department of Biology and Medical Biochemistry of CM UMK
Learning outcomes defined for a	Lecture:
given form of classes within the	W1: knows the structure of amino acids and their physical and
course	chemical properties - B K_W10
	W2: knows the structure of carbohydrates and their physical and chemical properties $PK_{ij}W10$
	and chemical properties - $B K_w 10$
	w5: knows the structure of lipids and their function at the collular and extracollular levels BK W11
	W4: knows the structure of polysaccharides and their function
	at the cellular and extracellular levels - B K W11
	W5: characterizes I-, II-, III- and IV structures of proteins- B K W12
	W6: knows the post-translational and functional modifications of the protein and their importance RK W12
	W7: characterizes proteins participating in chromatin structure
	- B K_W13
	W8: describes the role of enzymes and vitamins in metabolic processes - B K W15
	W9:knows the regulatory mechanisms of metabolic processes -
	BK_W15
	w10: describes the effect of proper diet, digestion processes and absorption on the course of metabolic pathways BK W15
	W11: describes antioxidant function of vitamins - B K W17
	W12: describes the role of blood in maintaining the body's
	water-electrolyte balance – B K_W01
	W13: characterizes the participation of blood buffers in
	maintaining body homeostasis $-BK_W02$
	changes on the properties of proteins and carbohydrates - B
	$\mathbf{\Lambda}_{-}$ UUS K1: is aware of the limitations of its knowledge and the need for
	its constant supplementing - K_K01
	Seminar:
	W1: knows the structure of amino acids and their physical and $P_{1} = P_{2} K_{1} W_{1}$
	<i>Chemical properties - B K_w10</i> <i>W2: knows the structure of carbohydrates and their physical</i>
	and chemical properties - B K_W10
	<i>W3: knows the structure of lipids and their function at the cellular and extracellular levels - B K W11</i>
	<i>W4: knows the structure of polysaccharides and their function</i>
	W5: characterizes I-, II-, III- and IV structures of proteins- B
	K_W12
	<i>wo: knows the post-translational and functional modifications</i> of the protein and their importance - <i>B K W1</i> ?
	<i>W7: characterizes proteins participating in chromatin structure</i>
	- BK_W13 W8: describes the role of ensures and vitaming in watch - lis
	processes - B K_W15

	W9:knows the regulatory mechanisms of metabolic processes -
	B K_W15
	W10: describes the effect of proper diet, digestion processes
	and absorption on the course of metabolic pathways - B K_W15
	W11: describes antioxidant function of vitamins - B K_W17
	W12: describes the role of blood in maintaining the body's
	water-electrolyte balance – B K_W01
	W13: characterizes the participation of blood buffers in
	maintaining body homeostasis $-B K_W02$
	K1: is aware of the limitations of its knowledge and the need for
	its constant supplementing - K_K01
	Tutorial:
	W1: knows the structure of amino acids and their physical and
	chemical properties - B K_W10
	W2: knows the structure of carbohydrates and their physical
	and chemical properties - B K_W10
	<i>W3: knows the structure of lipids and their function at the</i>
	cellular and extracellular levels - B K_W11
	<i>W4: knows the structure of polysaccharides and their function</i>
	at the cellular and extracellular levels - B K_W11
	W5: characterizes I-, II-, III- and IV structures of proteins- B
	K_W12
	W6: knows the post-translational and functional modifications
	of the protein and their importance - B K_W12
	W/: characterizes proteins participating in chromatin structure
	$-BK_W13$
	W6: describes the role of enzymes and vitamins in metabolic
	processes - D K_WIJ W0:knows the regulatory mechanisms of metabolic processes
	<i>R K W15</i>
	W10: describes the effect of proper diet digestion processes
	and absorption on the course of metabolic pathways - B K W15
	W11: describes antioxidant function of vitamins - BK W17
	W12: describes the role of blood in maintaining the body's
	water-electrolyte balance – B K_W01
	W13: characterizes the participation of blood buffers in
	maintaining body homeostasis $-B K_W02$
	U1: calculates molar and percentage concentrations of diluted
	solutions - B K_U03
	U2: determines the pH of the solution and the effect of pH
	changes on the properties of proteins and carbohydrates - B
	K_U05
	U3: uses qualitative analysis, titration, colorimetry, pH-
	parameters, analyzes and interprets the results obtained - B
	04: uses analytical balance, spectrophotometer, pH-meter and
	evaluates the accuracy of measurements $- B K_0 U to$
	K_1 . is aware of the initiations of its knowledge and the need for its constant supplementing - K K01
Assessment methods and criteria	Lecture:
for a given form of classes within	1. Intrasemestral test (MCO) (>60%): W1 W4-W13 1/2 K1
the source	2. Extended observation (>50%): K1.
	Seminar:
	1. Oral or written answer (>60%): W1-W13, K1.
	2. Oral presentation (>60%) :W1-W13, K1.
	3. Extended observation (>50%): K1.
	Tutorial:
	1. Intrasemestral test (MCQ) (>60%): W1, W4-W13, U1-U2,
	K1.
	2. Oral or written answer (>60%): W1-W13, K1.

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	5. Focused observation of the student's activity while		
	performing practical tasks (>60%): U1-U4.		
	4. Report (>60%): W1-W13, U1-U4, K1.		
	5. Extended observation (>50%): K1.		
	In the case of the intrasemestral test, the points obtained are converted into grades according to the following scale:		
	% points	Grada	1
	<i>76 points</i>	Uran and	-
	92-100	very gooa	-
	84-91	Good plus	-
	76-83	Good	_
	68-75	Satisfactory plus	
	56-67	Satisfactory	
	0-55	Fail	
	In the case of a learning outco criteria are ap - credit in the	an oral or written an. mes achieved by the plied: case when: the studer	swer to the assessment of the student, the following nt knows the basic issues and
	has mastered t	he program minimur	n understands the questions
	asked present	s his knowledge in a	logical and systematic way
	can practically	annly the acquired	knowledge:
	failure to page	s if: the student has	not mastered the program
	- juiure io pus	s ij. ine siudeni nus i s not understand the	austions, provides answers
	minimum, ades	s noi unaersiana ine	questions, provides answers
	not on the subj	ect, abes not properi	y use the basic vocabulary,
	can not practio	cally apply the acqui	red knowledge.
	In order to rec	eive the credit for the	e course the student must
	obtain positive	e results from two int	rasemestral tests and a
	positive assess	ment in the field of s	ocial competences (the
	appropriate ni	umber of points from	all assessed criteria).
Course content	Lectures:		
	1. Amino acids	s and peptides.	
	2 Structure an	d synthesis of protein	lS.
	3 Relationshi	ns hetween protein st	ructure and its function
	4 Blood functi	ions	actare and its function.
	5 Enzyma st	uns.	
	5. Enz,yme - sti	acture and propertie	
	0. Kinetics of e		
	7. water- ana	fat-soluble vitamins.	
	8. The role of e	enzymes in digestion	process.
	9. Classificatio	on of carbohydrates.	
	10. Lipid class	ification.	
	Seminars:		
	1. Disturbed st	tructure of proteins in	n the pathogenesis of selected
	diseases.		
	2. Enzymes in	diagnosis and treatm	ent.
	Tutorials:		
	1. Qualitative	analysis of amino ac	ids and proteins.
	2. \tilde{O} uantitative	e analysis of proteins	1
	3. Qualitative	and auantitative ana	lysis of blood components.
	4. Enzyme kind	etics.	,
	5 Qualitative	and auantitative and	lysis of selected vitamins
	6 Solocted pro	ma granuluive and	uicos
	7 Selected pro	peries of usesive j	abudratas
	7. Selected pro	pernes of monocarb	onyaraies.
The scheros and the l	o. <i>Kevision</i> .		
reaching methods	Lecture:	1	
	expository tea	ching methods - info	ormative (traditional) lecture
	with multimed	ia presentation, prob	olem-based teaching methods
	- discussion m	ethod	

	seminar: expository teaching methods - a talk, problem-based teaching methods - classic problem method, discussion method, case method, discovery teaching methods - project method tutorial: expository teaching methods - explanation, problem-based teaching methods – classic problem method, didactic discussion, discovery teaching methods - laboratory practicals, practical exercises, demonstration
Literature	The same as in part A